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The term Activation Energy was introduced in 1889 by Svante Arrhenius, a Swedish scientist. It is defined as the least possible amount of energy (minimum) which is required to start a reaction or the amount of energy available in a chemical system for a reaction to take place.

### **Definition of Activation Energy | Chegg.com**

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Si el sistema en el estado A se calienta isobáricamente (calentamiento sin cambio en la presión del sistema), entonces la humedad relativa del sistema disminuye porque la presión de vapor de agua en equilibrio aumenta con la temperatura en aumento.

### **Humedad relativa - Wikipedia, la enciclopedia libre**

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